

N-(2,6-Dimethylphenyl)-2-{3-[4-(methylsulfonyl)phenyl]-1,2,4-oxadiazol-5-yl}phenoxyacetamide

Hai-Bo Wang,* Jun Yin and Zhi-Tao Xing

College of Science, Nanjing University of Technology, Xinmofan Road No. 5
Nanjing, Nanjing 210009, People's Republic of China
Correspondence e-mail: wanghaibo@njut.edu.cn

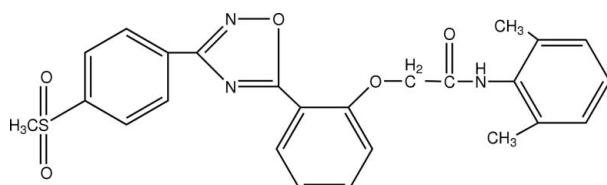
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Key indicators: single-crystal X-ray study; $T = 293\text{ K}$; mean $\sigma(\text{C}-\text{C}) = 0.007\text{ \AA}$;
 R factor = 0.075; wR factor = 0.178; data-to-parameter ratio = 16.4.

In the title compound, $\text{C}_{25}\text{H}_{23}\text{N}_3\text{O}_5\text{S}$, a bifurcated intramolecular $\text{N}-\text{H}\cdots\text{O}_2\text{N}$ hydrogen bond helps to establish the molecular conformation. The dihedral angles between the oxadiazole ring and its adjacent benzene rings are $24.50(19)$ and $12.20(18)^\circ$.

Related literature

For related literature, see: Romero (2001); Terashita *et al.* (2002).



Experimental

Crystal data

$\text{C}_{25}\text{H}_{23}\text{N}_3\text{O}_5\text{S}$
 $M_r = 477.52$

Monoclinic, $C2/c$
 $a = 28.848(6)\text{ \AA}$
 $b = 13.648(3)\text{ \AA}$
 $c = 13.562(3)\text{ \AA}$
 $\beta = 117.58(3)^\circ$

$V = 4733(2)\text{ \AA}^3$
 $Z = 8$
Mo $K\alpha$ radiation
 $\mu = 0.18\text{ mm}^{-1}$
 $T = 293(2)\text{ K}$
 $0.20 \times 0.10 \times 0.10\text{ mm}$

Data collection

Enraf–Nonius CAD-4
diffractometer
Absorption correction: ψ scan
(North *et al.*, 1968)
 $T_{\min} = 0.979$, $T_{\max} = 0.983$
4737 measured reflections

4643 independent reflections
3927 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.047$
3 standard reflections
every 200 reflections
intensity decay: none

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.075$
 $wR(F^2) = 0.178$
 $S = 0.97$
4643 reflections
283 parameters

146 restraints
H-atom parameters constrained
 $\Delta\rho_{\max} = 0.34\text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -0.39\text{ e \AA}^{-3}$

Table 1
Hydrogen-bond geometry (\AA , $^\circ$).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
N3—H3A \cdots O4	0.86	2.14	2.556 (5)	109
N3—H3A \cdots N2	0.86	2.43	3.288 (5)	176
C3—H3B \cdots O2	0.93	2.59	2.946 (7)	103
C6—H6A \cdots N2	0.93	2.56	2.901 (5)	102
C14—H14A \cdots O2 ⁱ	0.93	2.54	3.429 (7)	160
C16—H16A \cdots O2 ⁱ	0.97	2.57	3.310 (6)	133
C25—H25A \cdots N3	0.96	2.41	2.883 (6)	110

Symmetry code: (i) $x, y + 1, z$.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1996); software used to prepare material for publication: *PLATON* (Spek, 2003).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB2487).

References

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supplementary materials

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N-(2,6-Dimethylphenyl)-2-(2-{3-[4-(methylsulfonyl)phenyl]-1,2,4-oxadiazol-5-yl}phenoxy)acetamide

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Comment

1,2,4-Oxadiazole derivatives possess biological properties such as intrinsic analgesic (Terashita *et al.*, 2002) and antipicornaviral (Romero, 2001) effects. As part of our studies in this area, we report here the synthesis and crystal structure of the title compound, (I), (Fig. 1).

The dihedral angles between the N1/O1/C9/N2/C8 ring and its adjacent benzene rings are 24.50 (19) and 12.20 (18) $^{\circ}$ for the C2 and C10 rings, respectively.

An intramolecular, bifurcated N—H \cdots (N,O) hydrogen bond (Table 1) helps to establish the molecular conformation of (I). Some short C—H \cdots O and C—H \cdots N contacts are also present.

Experimental

2-Chloro-N-(2,6-dimethylphenyl)acetamide (10 mmol) was dissolved in acetone (100 ml) and potassium carbonate (15 mmol) was added. Then, 5-(2-hydroxyphenyl)-3-(4-methylsulfonyl)-phenyl-1,2,4-oxadiazole (10 mmol) was added to the reaction. The resulting mixture was refluxed for 12 h. After cooling and filtering, the crude title compound was obtained and purified by recrystallization from ethyl acetate. Crystals of (I) suitable for X-ray diffraction were obtained by slow evaporation of an ethanol solution.

Refinement

All H atoms were placed geometrically (N—H = 0.86 Å, C—H = 0.93–0.96 Å) and refined as riding with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{carrier})$ or $1.5U_{\text{eq}}(\text{methyl carrier})$.

Figures

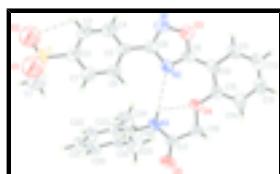


Fig. 1. A view of the molecular structure of (I) showing 30% displacement ellipsoids (arbitrary spheres for the H atoms). Dashed lines indicate the hydrogen bonds.

N-(2,6-Dimethylphenyl)-2-(2-{3-[4-(methylsulfonyl)phenyl]-1,2,4-oxadiazol-5-yl}phenoxy)acetamide

Crystal data

C₂₅H₂₃N₃O₅S

$F_{000} = 2000$

supplementary materials

$M_r = 477.52$	$D_x = 1.340 \text{ Mg m}^{-3}$
Monoclinic, $C2/c$	Mo $K\alpha$ radiation
Hall symbol: -C 2yc	$\lambda = 0.71073 \text{ \AA}$
$a = 28.848 (6) \text{ \AA}$	Cell parameters from 25 reflections
$b = 13.648 (3) \text{ \AA}$	$\theta = 9\text{--}12^\circ$
$c = 13.562 (3) \text{ \AA}$	$\mu = 0.18 \text{ mm}^{-1}$
$\beta = 117.58 (3)^\circ$	$T = 293 (2) \text{ K}$
$V = 4733 (2) \text{ \AA}^3$	Block, colourless
$Z = 8$	$0.20 \times 0.10 \times 0.10 \text{ mm}$

Data collection

Enraf-Nonius CAD-4 diffractometer	$R_{\text{int}} = 0.047$
Radiation source: fine-focus sealed tube	$\theta_{\text{max}} = 26.0^\circ$
Monochromator: graphite	$\theta_{\text{min}} = 1.6^\circ$
$T = 293(2) \text{ K}$	$h = -35 \rightarrow 31$
$\omega/2\theta$ scans	$k = 0 \rightarrow 16$
Absorption correction: ψ scan (North <i>et al.</i> , 1968)	$l = 0 \rightarrow 16$
$T_{\text{min}} = 0.979, T_{\text{max}} = 0.983$	3 standard reflections
4737 measured reflections	every 200 reflections
4643 independent reflections	intensity decay: none
3927 reflections with $I > 2\sigma(I)$	

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.075$	H-atom parameters constrained
$wR(F^2) = 0.178$	$w = 1/[\sigma^2(F_o^2) + (0.06P)^2 + P]$ where $P = (F_o^2 + 2F_c^2)/3$
$S = 0.97$	$(\Delta/\sigma)_{\text{max}} < 0.001$
4643 reflections	$\Delta\rho_{\text{max}} = 0.34 \text{ e \AA}^{-3}$
283 parameters	$\Delta\rho_{\text{min}} = -0.39 \text{ e \AA}^{-3}$
146 restraints	Extinction correction: none
Primary atom site location: structure-invariant direct methods	

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\text{sigma}(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S	0.14194 (6)	0.18555 (9)	0.81078 (14)	0.0925 (5)
O1	0.16502 (17)	0.1933 (3)	0.9251 (4)	0.126
O2	0.11504 (16)	0.1004 (3)	0.7585 (4)	0.125
O3	-0.06215 (11)	0.6306 (2)	0.5510 (2)	0.060
O4	0.06199 (10)	0.8113 (2)	0.6481 (3)	0.0673 (9)
O5	0.19856 (11)	0.8304 (2)	0.7637 (3)	0.0934 (13)
N1	-0.04744 (14)	0.5304 (3)	0.5690 (3)	0.0643 (11)
N2	0.02244 (12)	0.6304 (2)	0.6536 (3)	0.0453 (9)
N3	0.14383 (11)	0.7053 (2)	0.7465 (3)	0.0481 (9)
H3A	0.1118	0.6884	0.7242	0.058*
C1	0.1908 (2)	0.2025 (4)	0.7674 (5)	0.108
H1C	0.2148	0.1485	0.7927	0.163*
H1D	0.2094	0.2624	0.7983	0.163*
H1E	0.1747	0.2059	0.6877	0.163*
C2	0.10088 (18)	0.2876 (3)	0.7563 (4)	0.0582 (12)
C3	0.04852 (19)	0.2760 (3)	0.6783 (4)	0.0619 (13)
H3B	0.0351	0.2137	0.6536	0.074*
C4	0.01671 (17)	0.3567 (3)	0.6378 (4)	0.0615 (12)
H4A	-0.0183	0.3489	0.5867	0.074*
C5	0.03657 (15)	0.4475 (3)	0.6726 (3)	0.0446 (10)
C6	0.08838 (15)	0.4603 (3)	0.7502 (3)	0.0514 (11)
H6A	0.1015	0.5231	0.7731	0.062*
C7	0.12050 (17)	0.3807 (3)	0.7934 (4)	0.0640 (13)
H7A	0.1551	0.3891	0.8469	0.077*
C8	0.00352 (17)	0.5356 (3)	0.6299 (3)	0.0468 (10)
C9	-0.01894 (15)	0.6851 (3)	0.6063 (3)	0.0463 (10)
C10	-0.02598 (14)	0.7885 (3)	0.5990 (3)	0.0460 (10)
C11	-0.07551 (15)	0.8299 (3)	0.5674 (4)	0.0607 (12)
H11A	-0.1043	0.7892	0.5476	0.073*
C12	-0.08211 (19)	0.9311 (4)	0.5653 (4)	0.0719 (14)
H12A	-0.1153	0.9575	0.5423	0.086*
C13	-0.04122 (19)	0.9897 (3)	0.5960 (4)	0.0657 (13)
H13A	-0.0456	1.0569	0.5992	0.079*
C14	0.00762 (17)	0.9529 (3)	0.6232 (3)	0.0576 (12)
H14A	0.0354	0.9958	0.6411	0.069*
C15	0.01587 (15)	0.8543 (3)	0.6243 (3)	0.0459 (10)
C16	0.10881 (14)	0.8666 (3)	0.6885 (4)	0.0538 (11)
H16A	0.1104	0.9126	0.7446	0.065*
H16B	0.1095	0.9037	0.6281	0.065*
C17	0.15441 (16)	0.7993 (3)	0.7373 (4)	0.0529 (12)

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C18	0.18426 (14)	0.6307 (3)	0.7926 (3)	0.0416 (9)
C19	0.20946 (14)	0.6160 (3)	0.9060 (3)	0.0478 (10)
C20	0.24727 (16)	0.5436 (3)	0.9441 (4)	0.0659 (13)
H20A	0.2655	0.5334	1.0205	0.079*
C21	0.25960 (16)	0.4870 (3)	0.8786 (4)	0.0641 (13)
H21A	0.2854	0.4392	0.9087	0.077*
C22	0.23276 (17)	0.5023 (3)	0.7657 (4)	0.0633 (12)
H22A	0.2398	0.4620	0.7189	0.076*
C23	0.19522 (15)	0.5760 (3)	0.7186 (4)	0.0502 (11)
C24	0.16793 (17)	0.5935 (4)	0.5952 (4)	0.0819 (16)
H24A	0.1800	0.5468	0.5593	0.123*
H24B	0.1754	0.6586	0.5799	0.123*
H24C	0.1309	0.5862	0.5676	0.123*
C25	0.19565 (17)	0.6755 (3)	0.9839 (4)	0.0660 (13)
H25A	0.1686	0.7216	0.9413	0.099*
H25B	0.2261	0.7102	1.0363	0.099*
H25C	0.1836	0.6323	1.0231	0.099*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S	0.1070 (12)	0.0509 (8)	0.0930 (12)	0.0056 (8)	0.0237 (10)	-0.0028 (8)
O1	0.126	0.126	0.126	0.000	0.059	0.000
O2	0.125	0.125	0.125	0.000	0.058	0.000
O3	0.060	0.060	0.060	0.000	0.028	0.000
O4	0.0303 (15)	0.0592 (18)	0.104 (3)	0.0045 (14)	0.0243 (17)	0.0019 (18)
O5	0.0352 (18)	0.081 (2)	0.139 (3)	-0.0110 (16)	0.019 (2)	0.033 (2)
N1	0.045 (2)	0.066 (3)	0.079 (3)	-0.0122 (19)	0.026 (2)	-0.003 (2)
N2	0.0366 (19)	0.051 (2)	0.047 (2)	0.0039 (17)	0.0183 (17)	0.0049 (17)
N3	0.0237 (17)	0.054 (2)	0.063 (2)	0.0083 (15)	0.0174 (17)	0.0154 (18)
C1	0.108	0.108	0.108	0.000	0.050	0.000
C2	0.071 (3)	0.044 (2)	0.060 (3)	-0.011 (2)	0.031 (3)	-0.008 (2)
C3	0.082 (3)	0.039 (2)	0.064 (3)	-0.017 (2)	0.033 (3)	-0.010 (2)
C4	0.064 (3)	0.064 (3)	0.048 (3)	-0.016 (2)	0.019 (2)	-0.006 (2)
C5	0.045 (2)	0.054 (2)	0.039 (3)	-0.013 (2)	0.023 (2)	0.001 (2)
C6	0.049 (2)	0.049 (2)	0.054 (3)	-0.006 (2)	0.022 (2)	-0.007 (2)
C7	0.062 (3)	0.049 (2)	0.068 (3)	-0.005 (2)	0.019 (2)	0.002 (2)
C8	0.052 (3)	0.053 (3)	0.044 (3)	-0.005 (2)	0.029 (2)	0.001 (2)
C9	0.036 (2)	0.061 (3)	0.047 (3)	-0.011 (2)	0.024 (2)	-0.003 (2)
C10	0.041 (2)	0.063 (3)	0.032 (2)	0.0101 (19)	0.015 (2)	0.009 (2)
C11	0.039 (2)	0.072 (3)	0.067 (3)	0.013 (2)	0.021 (2)	0.009 (3)
C12	0.066 (3)	0.077 (3)	0.070 (4)	0.026 (2)	0.030 (3)	0.010 (3)
C13	0.065 (3)	0.056 (3)	0.069 (3)	0.022 (2)	0.025 (3)	0.020 (2)
C14	0.064 (3)	0.059 (3)	0.051 (3)	0.017 (2)	0.028 (2)	0.013 (2)
C15	0.044 (2)	0.054 (2)	0.034 (2)	0.0106 (19)	0.013 (2)	0.003 (2)
C16	0.043 (3)	0.056 (3)	0.054 (3)	-0.010 (2)	0.016 (2)	0.009 (2)
C17	0.043 (3)	0.042 (2)	0.056 (3)	-0.009 (2)	0.007 (2)	0.014 (2)
C18	0.032 (2)	0.043 (2)	0.046 (3)	-0.0009 (17)	0.0154 (19)	0.003 (2)

C19	0.039 (2)	0.057 (2)	0.044 (3)	0.0027 (19)	0.017 (2)	0.008 (2)
C20	0.049 (3)	0.074 (3)	0.065 (3)	0.008 (2)	0.018 (2)	0.014 (3)
C21	0.045 (3)	0.061 (3)	0.077 (3)	0.010 (2)	0.020 (3)	0.009 (3)
C22	0.062 (3)	0.060 (3)	0.075 (3)	0.001 (2)	0.038 (3)	-0.007 (3)
C23	0.044 (2)	0.062 (3)	0.049 (3)	-0.009 (2)	0.026 (2)	0.001 (2)
C24	0.065 (3)	0.108 (4)	0.060 (4)	0.013 (3)	0.018 (3)	-0.010 (3)
C25	0.073 (3)	0.070 (3)	0.069 (3)	0.000 (3)	0.045 (3)	0.002 (3)

Geometric parameters (\AA , $^\circ$)

S—O1	1.379 (5)	C10—C11	1.407 (5)
S—O2	1.395 (4)	C10—C15	1.413 (5)
S—C2	1.754 (4)	C11—C12	1.392 (6)
S—C1	1.777 (5)	C11—H11A	0.9300
O3—C9	1.343 (4)	C12—C13	1.322 (6)
O3—N1	1.420 (4)	C12—H12A	0.9300
O4—C15	1.349 (4)	C13—C14	1.375 (5)
O4—C16	1.417 (4)	C13—H13A	0.9300
O5—C17	1.228 (4)	C14—C15	1.366 (5)
N1—C8	1.313 (5)	C14—H14A	0.9300
N2—C9	1.298 (4)	C16—C17	1.486 (5)
N2—C8	1.383 (5)	C16—H16A	0.9700
N3—C17	1.337 (4)	C16—H16B	0.9700
N3—C18	1.453 (4)	C18—C19	1.378 (5)
N3—H3A	0.8600	C18—C23	1.399 (5)
C1—H1C	0.9600	C19—C20	1.382 (5)
C1—H1D	0.9600	C19—C25	1.524 (5)
C1—H1E	0.9600	C20—C21	1.343 (6)
C2—C7	1.388 (5)	C20—H20A	0.9300
C2—C3	1.395 (6)	C21—C22	1.375 (6)
C3—C4	1.374 (6)	C21—H21A	0.9300
C3—H3B	0.9300	C22—C23	1.398 (5)
C4—C5	1.356 (5)	C22—H22A	0.9300
C4—H4A	0.9300	C23—C24	1.502 (6)
C5—C6	1.384 (5)	C24—H24A	0.9600
C5—C8	1.476 (5)	C24—H24B	0.9600
C6—C7	1.371 (5)	C24—H24C	0.9600
C6—H6A	0.9300	C25—H25A	0.9600
C7—H7A	0.9300	C25—H25B	0.9600
C9—C10	1.422 (5)	C25—H25C	0.9600
O1—S—O2	120.0 (3)	C13—C12—H12A	120.0
O1—S—C2	107.0 (2)	C11—C12—H12A	120.0
O2—S—C2	109.8 (2)	C12—C13—C14	121.2 (4)
O1—S—C1	108.6 (3)	C12—C13—H13A	119.4
O2—S—C1	106.3 (3)	C14—C13—H13A	119.4
C2—S—C1	104.0 (2)	C15—C14—C13	120.9 (4)
C9—O3—N1	108.2 (3)	C15—C14—H14A	119.5
C15—O4—C16	121.3 (3)	C13—C14—H14A	119.5
C8—N1—O3	102.4 (3)	O4—C15—C14	125.3 (4)

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C9—N2—C8	104.5 (3)	O4—C15—C10	114.8 (3)
C17—N3—C18	122.6 (3)	C14—C15—C10	119.9 (4)
C17—N3—H3A	118.7	O4—C16—C17	109.4 (3)
C18—N3—H3A	118.7	O4—C16—H16A	109.8
S—C1—H1C	109.5	C17—C16—H16A	109.8
S—C1—H1D	109.5	O4—C16—H16B	109.8
H1C—C1—H1D	109.5	C17—C16—H16B	109.8
S—C1—H1E	109.5	H16A—C16—H16B	108.3
H1C—C1—H1E	109.5	O5—C17—N3	123.7 (4)
H1D—C1—H1E	109.5	O5—C17—C16	120.1 (3)
C7—C2—C3	119.7 (4)	N3—C17—C16	116.2 (3)
C7—C2—S	119.5 (3)	C19—C18—C23	123.0 (4)
C3—C2—S	120.8 (3)	C19—C18—N3	119.1 (4)
C4—C3—C2	120.1 (4)	C23—C18—N3	117.9 (4)
C4—C3—H3B	120.0	C18—C19—C20	116.0 (4)
C2—C3—H3B	120.0	C18—C19—C25	121.6 (4)
C5—C4—C3	119.8 (4)	C20—C19—C25	122.4 (4)
C5—C4—H4A	120.1	C21—C20—C19	124.7 (5)
C3—C4—H4A	120.1	C21—C20—H20A	117.7
C4—C5—C6	120.9 (4)	C19—C20—H20A	117.7
C4—C5—C8	121.1 (4)	C20—C21—C22	117.7 (4)
C6—C5—C8	118.0 (4)	C20—C21—H21A	121.2
C7—C6—C5	120.2 (4)	C22—C21—H21A	121.2
C7—C6—H6A	119.9	C21—C22—C23	122.3 (4)
C5—C6—H6A	119.9	C21—C22—H22A	118.8
C6—C7—C2	119.3 (4)	C23—C22—H22A	118.8
C6—C7—H7A	120.3	C22—C23—C18	116.2 (4)
C2—C7—H7A	120.3	C22—C23—C24	121.2 (4)
N1—C8—N2	113.8 (4)	C18—C23—C24	122.6 (4)
N1—C8—C5	122.3 (4)	C23—C24—H24A	109.5
N2—C8—C5	123.8 (4)	C23—C24—H24B	109.5
N2—C9—O3	111.1 (3)	H24A—C24—H24B	109.5
N2—C9—C10	132.4 (4)	C23—C24—H24C	109.5
O3—C9—C10	116.4 (4)	H24A—C24—H24C	109.5
C11—C10—C15	116.8 (4)	H24B—C24—H24C	109.5
C11—C10—C9	120.8 (4)	C19—C25—H25A	109.5
C15—C10—C9	122.3 (3)	C19—C25—H25B	109.5
C12—C11—C10	121.0 (4)	H25A—C25—H25B	109.5
C12—C11—H11A	119.5	C19—C25—H25C	109.5
C10—C11—H11A	119.5	H25A—C25—H25C	109.5
C13—C12—C11	119.9 (4)	H25B—C25—H25C	109.5
C9—O3—N1—C8	-2.3 (4)	C9—C10—C11—C12	176.9 (4)
O1—S—C2—C7	50.9 (4)	C10—C11—C12—C13	-1.7 (7)
O2—S—C2—C7	-177.3 (4)	C11—C12—C13—C14	4.4 (8)
C1—S—C2—C7	-63.9 (4)	C12—C13—C14—C15	-3.1 (7)
O1—S—C2—C3	-127.6 (4)	C16—O4—C15—C14	8.1 (6)
O2—S—C2—C3	4.2 (5)	C16—O4—C15—C10	-172.0 (4)
C1—S—C2—C3	117.6 (4)	C13—C14—C15—O4	178.8 (4)
C7—C2—C3—C4	0.5 (7)	C13—C14—C15—C10	-1.0 (6)

S—C2—C3—C4	179.0 (4)	C11—C10—C15—O4	-176.4 (4)
C2—C3—C4—C5	1.1 (7)	C9—C10—C15—O4	4.5 (6)
C3—C4—C5—C6	-1.2 (7)	C11—C10—C15—C14	3.5 (6)
C3—C4—C5—C8	179.4 (4)	C9—C10—C15—C14	-175.6 (4)
C4—C5—C6—C7	-0.2 (6)	C15—O4—C16—C17	163.9 (4)
C8—C5—C6—C7	179.2 (4)	C18—N3—C17—O5	2.2 (7)
C5—C6—C7—C2	1.7 (7)	C18—N3—C17—C16	-180.0 (4)
C3—C2—C7—C6	-1.8 (7)	O4—C16—C17—O5	170.2 (4)
S—C2—C7—C6	179.6 (4)	O4—C16—C17—N3	-7.7 (5)
O3—N1—C8—N2	0.9 (4)	C17—N3—C18—C19	86.6 (5)
O3—N1—C8—C5	178.2 (3)	C17—N3—C18—C23	-94.3 (5)
C9—N2—C8—N1	0.9 (5)	C23—C18—C19—C20	0.6 (6)
C9—N2—C8—C5	-176.4 (4)	N3—C18—C19—C20	179.7 (3)
C4—C5—C8—N1	9.5 (6)	C23—C18—C19—C25	-178.6 (4)
C6—C5—C8—N1	-169.9 (4)	N3—C18—C19—C25	0.4 (5)
C4—C5—C8—N2	-173.5 (4)	C18—C19—C20—C21	-1.6 (6)
C6—C5—C8—N2	7.1 (6)	C25—C19—C20—C21	177.7 (4)
C8—N2—C9—O3	-2.4 (4)	C19—C20—C21—C22	0.1 (7)
C8—N2—C9—C10	-179.3 (4)	C20—C21—C22—C23	2.5 (7)
N1—O3—C9—N2	3.1 (4)	C21—C22—C23—C18	-3.3 (6)
N1—O3—C9—C10	-179.5 (3)	C21—C22—C23—C24	177.7 (4)
N2—C9—C10—C11	-163.6 (4)	C19—C18—C23—C22	1.7 (6)
O3—C9—C10—C11	19.6 (6)	N3—C18—C23—C22	-177.4 (3)
N2—C9—C10—C15	15.4 (7)	C19—C18—C23—C24	-179.3 (4)
O3—C9—C10—C15	-161.4 (4)	N3—C18—C23—C24	1.6 (6)
C15—C10—C11—C12	-2.2 (6)		

Hydrogen-bond geometry (Å, °)

D—H···A	D—H	H···A	D···A	D—H···A
N3—H3A···O4	0.86	2.14	2.556 (5)	109
N3—H3A···N2	0.86	2.43	3.288 (5)	176
C3—H3B···O2	0.93	2.59	2.946 (7)	103
C6—H6A···N2	0.93	2.56	2.901 (5)	102
C14—H14A···O2 ⁱ	0.93	2.54	3.429 (7)	160
C16—H16A···O2 ⁱ	0.97	2.57	3.310 (6)	133
C25—H25A···N3	0.96	2.41	2.883 (6)	110

Symmetry codes: (i) $x, y+1, z$.

supplementary materials

Fig. 1

